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2	southern Italy)
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16	Abstract
17	The flux (Φ) of mercury (Hg) at the sediment-seawater interface was investigated in Augusta Bay
18	(southern Italy) where uncontrolled industrial discharge from one of the most important chlor-alkali
19	plant in Europe has caused significant negative effects on the environment. Hg fluxes were
20	measured by the deployment of in-situ benthic chamber. The obtained value of 1.3 kmol y ⁻¹ clearly
21	emphasizes the role of the sediments as source of Hg for the overlying water column. Moreover, Hg
22	concentrations in the outflowing bottom waters were measured to estimate the export of this
23	pollutant from Augusta Bay to the open sea. The calculated value of 0.54 kmol y ⁻¹ , corresponding to
24	~4% of the anthropogenic input of Hg from coastal point/diffuse sources to the Mediterranean Sea
25	(12.5 kmol y ⁻¹ ; Rajar et al. 2007; UNEP-MAP 2001), assigns this area a crucial role in the Hg
26	inventory of the entire Mediterranean basin. Finally, a consistent and robust mass balance for Hg in
27	Augusta Bay was provided by combining the obtained data with Hg fluxes at seawater-atmosphere
28	interface.

29 Keywords: mercury pollution, benthic fluxes, mass balance, Mediterranean Sea.

30 1. Introduction

31 The need to monitor mercury in coastal marine environments is an issue of great concern, especially where the anthropogenic impact is significant. Coastal areas play a key role in the global cycle of 32 this element (Mason et al., 1994), not only as a natural sink for terrestrially-derived Hg (Whalin et 33 al., 2007), but also as a potential source of methylmercury in the ocean (Mason and Benoit, 2003). 34 Despite the crucial importance of the coastal and shelf areas in the Hg biogeochemical cycle, few 35 papers have focused on the role played by these regions as pollutant sources for the open sea, 36 minimizing the role of Hg outflows in performing oceanic-scale mass balance (e.g., Cossa and 37 Coquery, 2004; Rajar et al., 2007). 38

Augusta Bay, situated on the eastern Sicily (Southern Italy), is one of the most polluted area in the 39 Mediterranean Sea, because of uncontrolled discharges (since1950s) from petrochemical plants. In 40 particular, the southernmost part of Augusta Bay hosted, from 1958 to 2005, an important mercury-41 cell chlor-alkali plant (Le Donne and Ciafani, 2008) which discharged without treatments in the 42 bay, until restrictions were imposed by Italian law in the late 1970s (Bellucci et al., 2012). 43 Documented effects of this indiscriminate discharges include: extremely high Hg concentrations in 44 the bottom sediment (up to 770 mgkg⁻¹) (ICRAM, 2008; Environ, 2008; Sprovieri et al., 2011), 45 significant Hg evasion fluxes to atmosphere (~10 g d^{-1}) (Bagnato et al., 2013) and a potential risk 46 associated with local fish consumption (Ausili et al., 2008; Bonsignore et al., 2013). Furthermore, 47 Bonsignore et al. (2015) measured Hg isotopes to trace transfer mechanisms of this toxic element 48 from sediment to the fish compartment and eventually to the resident population in Augusta area, 49 for which alarming increases in congenital malformations, abortions and mortality rates were 50 recorded (Madeddu et al., 2003). Recently, Sprovieri et al. (2011) used indirect calculations to 51 implement a Hg mass balance for the polluted Augusta Bay and estimated an outflow of ~0.16 52 Kmol y^{-1} to the open sea, which corresponds to ~1% of the Hg anthropogenic input to 53

54 Mediterranean sea from point and diffuse sources (12.5 Kmol y^{-1} ; Rajar et al., 2007). This result 55 emphasizes the role played by Augusta Bay as source of mercury at Mediterranean scale.

Here, we present a revised and more quantitatively robust mass balance of Hg in Augusta Bay 56 based on unprecedented information obtained by benthic chamber experiments, measurements of 57 Hg in outflowing seawater and evasional fluxes to atmosphere (Bagnato et al., 2013). The study 58 offers the valuable opportunity to provide a comprehensive vision of the geochemical cycle of Hg 59 in Augusta Bay, with particular attention paid to the benthic fluxes. It also explores the role played 60 by Augusta bay as potential Hg point source for the Mediterranean sea. In fact, owing to its 61 geographical location, the outflowing shelfwater is immediately intercepted by the surface Atlantic 62 Ionian Stream (AIS) and mixed with the main gyres of the eastern Mediterranean Sea, thus 63 representing a risk in terms of large-scale marine system. 64

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66 2. Study area

Augusta Bay is a natural semi-enclosed marine area of ~23.5 km², located in Eastern Sicily (Ionian 67 Sea, southern Italy), delimited in the northern sector by the town of Augusta and closed to South 68 and East by artificial dams built in the early '60s (Fig. 1). It hosts one of the most important harbour 69 of the Mediterranean Sea characterized by an intensive ship traffic. Two main mouths allow 70 connection with the open sea: Scirocco (300 m wide and 13 m deep) and Levante inlets (400 m 71 wide and 40 m deep). The exchanges with open sea are mainly driven by tidal fluctuations and, 72 consequently correlated with the entry/exit of tidal flows and relative amplitudes. The input water at 73 Levante mouth is characterized by a mean speed of 18 cm s⁻¹ (depth= 0.5 m) at the surface and 7 74 cm s⁻¹ at the bottom. (depth= 30-40 m). The input water circulation flows northward, parallel to the 75 dam, while, the output current flows in opposite directions, with speeds of 5-6 cm s⁻¹ (depth= 5-30) 76 m) The Scirocco mouth is mainly affected by outflowing water, which goes parallel to the coastline 77 with moderate speeds (5-6 cm s⁻¹). Finally, the northern part of the bay is scarcely affected by active 78

currents (ICRAM, 2008; Sprovieri et al., 2011). Recent bathymetric surveys carried out by Budillon 79 et al. (2008), showed a very narrow shelf develops down to 100-130 m with a mean gradient of 80 about 1.0° and a steep slope characterized by a dense net of canyons dropping to the deep end of the 81 Ionian basin (Fig. 1). As consequence of the industrial development, which has affected the area 82 since 1960s, a wide range of pollutants has been introduced into Augusta Bay. In particular, high 83 levels of Hg contamination resulted by the activities of the chlor-alkali plant, which discharged 84 without treatments until the 1970s, when a demercurization plant and waste treatments became 85 operative (Bellucci et al., 2011). Owing to the high state of environmental degradation, this area was 86 included in 2002 in the National Remediation Plan by the Italian Environmental Ministry. 87

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3. Materials and sampling strategy

The sampling strategy included the collection of seawater and sediment, as well as the placement of 90 a benthic chamber at the bottom of the bay to estimate the Hg fluxes at the sediment-seawater 91 interface (Fig. 1, Tab. 1). In May 2011, seawater samples were collected from the whole study-area 92 (Fig. 1a), while sediment sampling stations (Fig. 1b) were selected in order to cover the range of Hg 93 concentration previously reported from other studies (ICRAM, 2008; Environ, 2008; Sprovieri et 94 al., 2011). Hg distribution obtained from the analysis of these samples drove the benthic chamber 95 96 deployment performed in September 2011 and June 2012 (Fig. 1c), in order to measure fluxes from sediments characterized by different Hg pollution levels. In June 2012, also seawater and sediment 97 were sampled in the same stations where benthic chamber experiments were performed (Fig. 1 a, b, 98 c). Finally, in February 2012, seawater samples were collected outside the bay in order to 99 investigate the export of Hg to open sea (Fig. 1a). 100

101

102 3.1 Seawater collection

Seawaters were collected from 18 stations inside and 4 outside the bay, at three different depths
(surface, mid-water and bottom) with a Niskin bottle rosette or a single Niskin bottle transported by

a scuba diver (Fig. 1a; Tab. 1). All of the samples were preserved in bottles, previously cleaned with HNO₃/HCl (10%) and rinsed with Milli-Q water (18.2 M Ω cm⁻¹). Samples were stored at -20°C until the analyses (Horvat et al., 2003). Measurements of temperature and salinity along the water column were obtained by multiparametric probe (CTD SBE 9plus).

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110 3.2 Sediment sampling

Sediments were collected using a box-corer or a Plexiglas tube (length=30 cm; diameter=6 cm),
manually inserted into the bottom sediment by a scuba diver (Tab. 1; Fig. 1b) and immediately
stored at -20 °C until the analyses.

114

115 3.3 Benthic chamber

To calculate the Hg fluxes at the sediment-seawater interface, a benthic chamber was designed and 116 117 assembled at the IAMC-CNR-Capo Granitola, where the base model proposed by Covelli et al. (1999, 2008) was modified. Briefly, a Plexiglas box-shape chamber (50x50x30 cm), open at the 118 119 bottom side, was internally equipped with a stirring mechanism. This consisted of a rolling bar (30 cm long; 5 rpm speed) moved by an electromotor (12 V) located on the top of the chamber. A 120 plastic skirt fitted outside the chamber controls its penetration into the sediment to a depth of 7.5 121 cm. Seawater from the benthic chamber was periodically sampled at discrete, consecutive temporal 122 intervals (t=0, t= 1, t=4, t=6, t=10 and t=12 h) with a syringe inserted into a cap with a rubber 123 pierceable septum. It was immediately transferred into acid-pre-cleaned vials and stored at T= -124 20°C until analyses. 125

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127 **4. Analytical methods**

Analyses were performed at Geochemical Laboratory of IAMC-CNR - Capo Granitola. In order to
 minimize contamination, acid-cleaned laboratory materials and ultrapure grade reagents were used
 during sample preparation and analyses.

131

132 4.1 Seawater

Total (THg) and dissolved Hg (DHg) in the seawater were determined using a direct mercury 133 analyzer (Milestone_DMA-80) (US-EPA 7473), after oxidation with a bromine monochloride 134 (BrCl) fresh solution (0.5 mL/100 mL sample) (US-EPA 1631). The DHg was determined after 135 sample filtration (0.45 µm), in seawater samples collected from the benthic chamber and outside the 136 bay. Blank and duplicate samples (about 20% of the of samples) were analyzed to assess detection 137 limit (d.l.= 1.9 ng L^{-1} ; 3 σ of the reagent blank), and reproducibility (better than 10%) of the method. 138 Accuracy was estimated by spiked samples analysis being the certificated value of the available 139 Reference Standard Material (BCR-579 Coastal Seawater) 1-2 orders of magnitude lower than 140 investigated analytical range. Obtained recoveries resulted to be between 85 and 110%. 141

142

143 4.2 Sediment

Once defrosted, extruded from the liner, sediment cores were sectioned at 1-3 cm intervals with a 144 145 stainless steel band saw. About 3.5 g of wet sediment were picked from each slide for the grain size analyses, performed with a laser granulometer (Horiba Partica LA-950V2) after treatment with 146 H₂O₂ (2:8) and Milli-Q water to remove organic material and cemented salts (Romano and 147 Gabellini, 2001). The remaining sediment were dried at 35°C and powdered using an agata mortar. 148 About 10 mg of dry sediment were loaded into specific nickel boats and analyzed by DMA-80 for 149 Hg analysis (US-EPA 7473). A Reference Standard Material (PACS-2 Marine sediment, NRCC) 150 was analyzed to assess accuracy (estimated to be $\sim 7\%$) and precision (routinely better than 6%; 151 RSD%, n= 3). Finally, duplicate samples (about 20% of the total samples) were analyzed to 152 estimate the reproducibility (better than 7%). Total organic-carbon (TOC) was determined in bulk 153 sediment by a Thermo Electron Flash EA 1112 after elimination of all the carbonate (with HCl 1M 154 for 24 h at T_{ambient} and drying at 60°C) (Nieuwenhuize et al., 1994). 155

156

157 **5. Results**

158 5.1 Seawater

Total and dissolved mercury concentrations measured in the seawater sampled within and outside Augusta Bay are reported in Table 2. The [THg] in the internal seawater vary widely from <1.90 (<d.l.) to 129 ngL⁻¹ (Tab. 2), with an evident north-south increasing trend (Fig. 2).

162 The highest [THg] occurred in south-western area (S11-S25), ranging between 3.3 and 129 ngL⁻¹ in 163 the 10-20m depth interval, between 2.27 and 127 ngL⁻¹ in the deep waters (<10m from bottom) and 164 from <1.90 to 26.3 ngL⁻¹ in the shallow waters (>20m from bottom) (Fig. 2).

It is noteworthy that extremely high levels of THg were measured in S23, at the bottom ([THg]= 165 57.8 ngL⁻¹), and in S17, at middle and deep depths ([THg]= 129 and 127 ngL⁻¹ respectively) (Tab. 166 2). Lower concentrations (average= $12.0\pm7.00 \text{ ngL}^{-1}$) were found in the northern stations (S1-10) 167 (Tab. 2). Furthermore, an increasing trend of [THg] with depth was highlighted along the seawater 168 column, where, systematically, the surface waters show the lowest THg content (average= 169 11.1 \pm 8.49 ngL⁻¹), the deep waters have the highest one (average= 25.5 \pm 29.3 ngL⁻¹) and the middle 170 waters have intermediate THg levels (average= 20.1 ± 29.1 ngL⁻¹). The [THg] and [DHg] measured 171 in seawater samples outside the bay (S26-29) varied between 2.62 and 11.9 ngL^{-1} and between <d.1. 172 and 5.55 ngL⁻¹, respectively. Once again, an increasing trend of [THg] with depth was measured. In 173 particular, the highest values (THg= 11.0 and 11.9 ngL^{-1} ; DHg= 5.55 ngL^{-1}) were measured in the 174 bottom seawaters (Tab. 2). The range of temperature measured inside the bay is relatively wide 175 (from 16.6 to 20.4°C), with warmer surface water (T range= 19.0-20.4°C) and colder deep water (T 176 range= 16.6-20.2°C). The low temperatures observed in the eastern part of the bay (S25, S13) are 177 due to the input of relatively colder water from the open sea, with measured values ranging from 178 13.9 to 15.2 °C (Tab. 2). Salinity ranges from 36.7 to 39.0 psu inside and between 37.9 and 38.9 psu 179 outside the bay. The lower values recorded inside the bay are probably due to inputs of freshwater 180 or sewage from land (Tab. 2). 181

182

183 5.2 Sediment

Grain size, TOC and [Hg] measured in the sediment are reported in Figure 3 (a, b, c, respectively)and Appendix I.

Almost all of the analyzed samples consist of silt (more than 50%) and clay (32-45%), whereas the sandy fraction represents a small percentage (~10%). Only the S19 core is sandy (~70%) and contains small amounts of silt (~17%) and clay (~13%). The calculated sorting parameter (ranging from 4.8 to 12.2 μ m) is indicative of a low energy environment, where the slow current circulation affects the poor size selection of the sediment (Fig. 3a).

All samples show high TOC (from 1.58 to 4.26%), thus suggesting relatively high burial/preservation of organic matter in the sediment. In the southern S21, the values range widely from 1.58 to 3.45% (average= $2.32\pm0.50\%$) with an irregular distribution along the core, characterized by an evident peak (3.45%) at 14-16 cm. In the S15 core, the TOC-depth profile is quite constant except for two peaks at 2-4 cm (2.85%) and 20-22 cm (4.26%) (average= $2.38\pm0.68\%$). The lowest mean value (2.27\pm0.11%) was measured in the northern station (S7), where TOC show a steady trend with depth (Fig. 3b).

The measured [THg] varies widely (from 1.77 to 55.3 mgKg⁻¹) (Fig. 3b). The highest values were 198 found in the southern stations, specifically in the intermediate part (6-21 cm) of the core S14, with 199 [THg] ranging between 20.8 and 39.5 mgKg⁻¹, and along the core S21, where the values vary 200 between 19.0 and 55.3 mgKg⁻¹ in the interval from 4 to 30 cm). Moreover, the concentration-depth 201 profiles in cores S14 and S21 are characterized by a marked peak between 12 and 16 cm, whereas 202 nearly steady trends were measured in the remaining cores. The lowest values were found in the 203 northern stations, with average concentrations of 6.33±0.91 mgKg⁻¹ in S7, 6.15±1.54 mgKg⁻¹ in S8; 204 and in the sandy S19 core (average= 7.81 ± 1.66 mgKg⁻¹) collected from the southern area (average= 205 $7.81\pm1.66 \text{ mgKg}^{-1}$). Finally, intermediate values were measured in S15 (average= $12.5\pm2.66 \text{ mgKg}^{-1}$) 206 ¹) and S16 (average= $16.2 \pm 1.66 \text{ mgKg}^{-1}$) (Fig. 3c). 207

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209 5.3 Benthic fluxes

The DHg content in the seawater at the sediment-seawater interface increased with time at all stations (Appendix II, Fig. 4). Specifically, the increasing trend was constant for the entire incubation period (0-10/12h) in chambers 18 and 21, while in the other stations the DHg content exhibited a major increase in the first hour (0-1h), but was almost constant during the time interval that followed (1-10/12h) (Fig. 4). The mercury fluxes across the sediment-seawater interface were calculated on the basis of the best fit curve "concentration *vs* time", according to the following equation (Santschi et al., 1990) (1):

$$\phi Hg = \left(\frac{\delta C}{\delta t} \times \frac{V}{A}\right) \tag{1}$$

218 where:

219 $\delta C/\delta t$ = variation of the Hg concentration over time (ng L⁻¹ h⁻¹).

220 V= volume of the benthic chamber (57.5 L).

A = cover area of the sea-bottom (0.25 m²).

222

Finally, the obtained results were normalized to 24h. The mercury fluxes per day calculated in the 223 northern area were very comparable in the neighboring S7 and S9 (22.7 and 22.6 $\mu g~m^{-2}~d^{-1},$ 224 respectively) (Fig. 4). Conversely, in the southern area, significant differences among the stations 225 were observed . In particular, S15 and S22 had the highest values (92 and 56 $\mu g\ m^{-2}d^{-1}$, 226 respectively), while relatively low flux (8.7 μ g m⁻²d⁻¹) were estimated for S18, despite the closeness 227 to S21, where a higher flux of 20.5 μ g m⁻²d⁻¹ was recorded (Fig. 4). The calculated fluxes were 228 extended to a yearly estimation and then over the entire area using the territorial distribution model 229 proposed by Aurenhammer (1991) (Voronoi Polygons method). The estimated fluxes amounted to 230 1.1 kmol y⁻¹ in September 2011 (0.22 t y⁻¹) and 1.4 kmol y⁻¹ (0.29 t y⁻¹) in June 2012. 231

232

233 6. Discussion

The [Hg] measured in Augusta Bay sediment are extremely elevated (Fig. 3c), being 1-2 orders of 234 magnitude higher than the Mediterranean background values (0.04-0.17 mgKg⁻¹) (e.g., Donazzolo 235 et al., 1981; Ogrinc et al., 2007). They are also comparable with the values (4.06 and 79.0 mg Kg⁻¹) 236 reported for sediment of other coastal areas affected by relevant industrial or mining activities (e.g., 237 Baldi and Bergagli, 1984; Covelli et al., 2001, 2008, 2011; Emili et al., 2012). The higher 238 concentrations, measured in the southern stations, can be attributable to industrial input ($\sim 2.1 \text{ kg y}^-$ 239 ¹; value reported by the European Pollutant Emission Register) and to the proximity of the most 240 important chemical and petrochemical plants (Syndial Priolo Gargallo, ESSO, ERG, etc.). In 241 particular, the chlor-alkali plant, is considered to have been the main Hg source in the bay (Di 242 Leonardo et al., 2007; ICRAM, 2008; Sprovieri et al. 2011; Bellucci et al., 2012), with an average 243 discharge of about 260 kg y⁻¹ of Hg estimated between early-1960 and late-1970 (Colombo et al., 244 2005). However, in Augusta Bay, effects of re-suspension due to the intensive ship traffic, 245 combined with relevant dredging activities occurred since 1960s, make the historical reconstruction 246 of industrial contaminant inputs very difficult to be achieved without a specific integrated approach 247 (Cundy et al., 2003; Bellucci et al., 2012). Concentrations measured in the studied cores are in good 248 agreement with the Hg areal distribution in surface sediment reported by Sprovieri et al. (2011), 249 which subdivided Augusta Bay in two geographical sectors characterized by significantly different 250 levels of contamination: i) a southern sector ([Hg] median value=23.8 mgKg⁻¹) affected by chemical 251 and petrochemical plants activity and ii) a northern sector ([Hg] median value= 1.1 mg Kg⁻¹) 252 affected by municipal waste discharge. The only exception is represented by S19 core, 253 characterized by Hg contents lower than those found in the neighboring stations. It is probably due 254 to its specific sandy grain-size composition (Fig. 3a) responsible for a less Hg adsorbing capacity 255 (Ravichandran, 2004). On the other hand, a positive correlation was found between [Hg] and finer 256 grain-size fraction for S14 ($r^2 = 0.7$) and S21 ($r^2 = 0.5$), where specifically, the highest Hg levels 257 correspond to the highest silt-clay percentages (92 and 95%) (Fig. 3 c, a). The high TOC values 258

measured (Fig. 3, Appendix I) are compatible with the disoxic/anoxic conditions measured at the 259 bottom basin (average Eh~ -300 mV; ICRAM, 2008; Sprovieri et al., 2011) and possibly due to the 260 state of semi-closed environment of the basin and the low energy hydrodynamics. Noteworthy, high 261 contents of organic matter under reducing environmental conditions promote methylation processes, 262 mediated by microorganisms, such as sulfate reducing bacteria (SRB), at water/sediments interface 263 (e.g., Langer et al., 2001; Lambertsson and Nilsson 2006). Thus, bottom sediments in Augusta bay 264 represent a potential source of methylmercury (MeHg) for the ecosystem which could also partially 265 justify the high levels of Hg found in fish caught in the bay (Bonsignore et al., 2013, 2016). 266

A significant statistical correlation ($r^2 = 0.71$) between [THg] and TOC, was found in the core S21, 267 where peaks of Hg corresponding to the highest TOC values (Fig. 3 c, b) occur. Conversely, no 268 positive correlation was recorded in the other cores (S7, S15). The north-south positive gradient of 269 [Hg] in the Augusta sediment is reflected in the spatial distribution of THg in seawater, with lower 270 271 values recorded in the northern area and higher ones in the south (Fig. 2). Noteworthy, an evident [THg] vertical increasing trend was observed along the water column (Fig. 2, Tab. 2), with the 272 lowest concentrations measured in the surface layers, where the penetrating sunlight promotes the 273 photoreduction and the subsequent volatilization of Hg⁰ form (Fitzgerald and Clarkson, 1991; 274 Bagnato et., 2013), and the highest values found close to the bottom, where resuspension events and 275 Hg release from sediment probably affect the concentration of this pollutant in the overlying water 276 277 column. In the marine environment, Hg is generally bound to sediment through: ion exchange, complexing or chelation with organic and inorganic ligands, sorption with Fe/Mn oxide-hydroxides 278 or incorporation into mineral lattice (Ramamoorthy and Massalki, 1979). However, changes in 279 physical-chemical parameters at the sediment-seawater interface, bioturbation, dredging activities 280 and methylation, may result in remobilization and diffusion of the Hg in the water column (Covelli 281 282 et al., 1999). In this study, benthic chamber experiments provided direct measurement of Hg fluxes from the sediment to the overlying seawater. The obtained values vary widely between 8.7 and 92 283 µg m⁻²d⁻¹. These differences could be attributed to variable Hg concentrations in surface sediments 284

(Appendix I), where comparable physical-chemical conditions at the seawater-sediment interface 285 and analogue Hg speciation occurs, as well-demonstrated by Oliveri et al. (2016, in press). The 286 measured benthic fluxes are higher than those estimated for other contaminated areas (5 and 10 µg 287 m⁻² d⁻¹) (Bothner et al., 1980; Covelli et al., 1999; Boucher et al., 2011) and comparable with values 288 reported by Covelli et al. (2008) for the highly-contaminated Grado Lagoon (37 and 77 μ g m⁻² d⁻¹). 289 The flux at the sediment-seawater interface estimated for the entire bay $(1.3\pm0.1 \text{ kmol s}^{-1})$ is about 290 two orders of magnitude higher than that of the evasion into the atmosphere $(1.8 \times 10^{-2} \text{ kmol y}^{-1})$ 291 (Bagnato et al., 2013). This could imply that a relevant part of the Hg released in seawater is 292 recycled in the biogeochemical cycle (biota uptake, adsorption by suspended particulate matter, re-293 deposition) and/or outflows to the open sea. This latter hypothesis is supported by the anomalous 294 Hg concentrations measured in the seawater outside Augusta Bay (Tab. 2), which were an order of 295 magnitude higher than the Mediterranean background values (0.2-0.4 ng L^{-1}) (Cossa et al., 1997; 296 Horvat et al., 2003; Kotnik et al., 2007; Rajar et al., 2007). This is in agreement with the data 297 reported by Fantozzi et al., (2012), which measured unusual Hg contents in seawater out Augusta 298 Bay (4.8 and 18 ngL⁻¹) and argued that the water circulation in the Strait of Sicily moves polluted 299 masses from Augusta Bay eastwards. Moreover, Sprovieri et al. (2011) speculated that the effects of 300 the meso-scale circulation of the Ionian Sea, together with a narrow shelf and a steep slope off 301 Augusta Bay (Budillon et al., 2008) (Fig. 1), create a high potential risk of Hg contamination of the 302 Mediterranean basin. This evidence calls for an appropriate estimate of the possible outflow of this 303 pollutant from Augusta Bay to the open sea. To this aim, we took in account the DHg 304 concentrations measured in the bottom water just outside Scirocco and Levante mouths (5.6 and 3.6 305 ng L⁻¹, respectively; Tab. 2) and the associated relative outputs of seawater (2.34 x 10^{13} kg y⁻¹), as 306 reported by Sprovieri et al. (2011). Thus, the estimated Hg outflow resulted to be 0.54 kmol y^{-1} . 307 This value corresponds to ~4% of the anthropogenic input of Hg from coastal point/diffuse sources 308 to the Mediterranean Sea (12.5 kmol y⁻¹; Rajar et al. 2007; UNEP-MAP 2001), and definitively 309 emphasizes the role played by Augusta Bay as a crucial "point source" at the basin scale. 310

311

312 Mass balance of Hg in Augusta Bay

A steady-state mass balance for mercury, calculated to explore the geochemical cycle of this pollutant within the Augusta Bay, is here proposed. For this purpose, the Augusta Bay was basically considered as an "environmental compartment" with well-defined borders and constrained hydrodynamics. According to the basic scheme proposed by Sprovieri et al. (2011), we applied the following equation:

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319

I+A+AD+R = O+D+V

(2)

320 where:

I = total Hg influx from the surface Mediterranean seawater

322 A= input of dissolved Hg from anthropogenic activities

323 AD= atmospheric Hg deposition

R = Hg re-suspension/release from the sediment

O = Hg outflow from the basin to the open sea

326 D= amount of Hg recycled into Augusta Bay (buried in sediment and/or accumulated in the biota)

V = Hg evasion into the atmosphere

328

The total seawater Hg influx value (I) was estimated using data reported by Kotnik et al. (2007), and 329 corresponded to an average input of Hg from the Ionian Sea to Augusta Bay of about 3.12±0.94 x 330 10^{-2} kmol y⁻¹. The anthropogenic Hg input into the basin (A), including discharges from wastewater 331 treatment facilities and industrial activities (Erg, Syndyal, Esso) was considered to be 6.2 x 10⁻² 332 kmol y⁻¹ (value reported by the European pollutant emission register for the year 2005). AD and V 333 corresponded to the values reported by Bagnato et al. (2013) (0.42 x 10^{-2} kmol y⁻¹ and 1.7±0.02 x 334 10^{-2} kmol y⁻¹, respectively). Finally, O and R are the values measured in this study, which 335 correspond to the Hg outflow from the bay to the open sea (0.54 ± 0.08 kmol y⁻¹), and the Hg flux 336 from the sediment to the seawater $(1.3\pm0.2 \text{ kmol y}^{-1})$, respectively. In particular, the last term 337 represents the final result of Hg diffusion from porewater and desorption from re-suspended of 338

organic/inorganic particles. Thus, the Hg recycled (re-deposited and/or accumulated in the biota as MeHg) into the bay (D) results to be 0.84 ± 0.22 kmol y⁻¹, which represents from ~45 to 75% of the total input of Hg in the Augusta Bay. while the remaining part escapes into atmosphere (1-2%) and outflows to open sea (average ~40%). This means that the Augusta Bay plays an important role in recycling and exporting of mercury into the Mediterranean Sea and represents a relevant point source for that system.

This study definitively states how strong interlinks between contaminants behaviour and ocean dynamics drive the specific control of point sources on widespread distribution of pollutants. In particular, this offers a good science-driven case for an appropriate conceptual approach to the Marine Strategy Framework Directive (2008/56/EC), which specifically states that it should be possible to determine how to keep the pressure of human activities within levels compatible with the preservation or restoration of a Good Environmental Status in the Mediterranean sea, from 2020 onwards.

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359

360 Figure captions:

Figure 1: Sampling location of the seawater (a), sediment (b) and benthic chamber (c) in theAugusta Bay.

- **Figure 2:** Map of [THg] distribution in seawaters collected during May 2011; a) 0-10m from the
- bottom; b) 10-20m from the bottom; c) 20m from the bottom up to surface. Maps were created
- through3D block kriging and were themed on the "natural breaks" method.
- **Figure 3:** Profile of TOC (a), [Hg] (b) and grain size composition (b) along the collected cores.
- **Figure 4:** Concentration of DHg *vs*. time during the benthic chamber experiment.
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Sampling	Vessel	Period	Station			Bottom depth	Method of sampling	
				Lat.	Long.	(m)		
			1	37°14.392N	15°12.537E	14.0		
			2	37°13.864N	15°12.519E	16.0		
			3	37°13.863N	15°11.845E	11.0		
			4	37°13.353N	15°11.902E	8.00		
			5	37°12.849N	15°12.595E	20.0		
			6	37°12.743N	15°13.176E	17.0		
	N/O	22	10	37°12.267N	15°11.863E	22.0		
	N/U C. Dallanarta	23- 26/05/11	11	37°12.009N	15°12.161E	22.0	Niskin	
	G. Dallaporta	20/03/11	12	37°11.961N	15°12.855E	27.0		
S			13	37°11.943N	15°13.536E	32.0		
iter			17	37°11.696N	15°12.512E	24.0		
1W5			20	37°11.363N	15°13.536E	21.0		
Sea			23	37°11.445N	15°12.434E	22.0		
			24	37°11.434N	15°13.536E	21.0		
			25	37°10.871N	15°12.833E	16.0		
			26	37°10.310N	15°13.148E	17.0		
	N/O	02/02/12	27	37°11.807N	15°14.328E	42.0	Nielein	
	Urania	02/02/12	28	37°11.742N	15°15.767E	85.0	INISKIII	
			29	37°11.604N	15°18.240E	679		
	M/N	23	7	37°12.578N	15°12.583E	22.8		
	L. Sanzo	23- 26/06/12	15	37°11.697N	15°12.917E	27.8	Single Niskin bottle	
			21	37°11.288N	15°12.459E	22.8		
	N/O		8	37°12.618N	15°12.473E	22.0		
S	G. Dallaporta	23- 26/05/11	14	37°11.745N	15°12.985E	28.0	Box-corer	
ent			16	37°11.818N	15°12.540E	27.0	Box colei	
ime	Danaporta		19 37°11.399N 1	15°12.375E	21.0			
edi	M/N L. Sanzo	23-	7	37°12.578N	15°12.583E	22.8		
S		26/06/12	15	37°11.697N	15°12.917E	27.8	Manual corer	
		20,00,12	21	37°11.288N	15°12.459E	22.8		
	M/N L. Sanzo	19-	9	37°12.369N	15°12.289E	22.0		
r af er ce		21/09/11	18	37°11.294N	15°12.242E	21.0		
nten ner /ato			22	37°11.406N	15°12.572E	24.0	Benthic chamber	
uwa din eaw	M/N	M/N	23-	7	37°12.578N	15°12.583E	22.8	(syringe)
Sea se st in	L. Sanzo	26/06/12	15	37°11.697N	15°12.917E	27.8		
		20,00,12	21	37°11.288N	15°12.459E	22.8		

Table 1: Details of the sampling performed in Augusta Bay.

	Sampling			Denth	Т	<u>.</u> Salinity	THσ	DHa
	neriod	S	tation	(m)	(C°)	(<i>nsu</i>)	$(n \circ L^{-1})$	$(n \rho L^{-1})$
	periou		Rottom	11.2	18.0	38.0	17.8	(1181)
		1	Middle	6.20	18.4	38.0	9.17	
		-	Surface	1 40	20.4	38.0	9.17	
			Rottom	10.7	18.0	38.0	177	
		2	Middle	6 71	18.8	38.0	14.9	
		-	Surface	2 21	20.0	38.0	<d 1<="" td=""><td></td></d>	
			Rottom	8 40	18.1	38.0	30.0	
		3	Middle	4 60	19.7	38.0	12.0	
		v	Surface	2.26	20.3	38.0	<d 1<="" td=""><td></td></d>	
			Rottom	3.12	20.2	38.0	6 30	
		4	Surface	0.10	20.2	38.0	<d 1<="" td=""><td></td></d>	
			Rottom	15.9	17.7	38.0	17.7	
		5	Middle	9 20	18.5	38.0	7 10	
		U	Surface	1.00	19.8	38.0	9 20	
			Rottom	13.5	18.2	38.2	20.6	
		6	Middle	6 74	19.0	37.6	3 40	
		v	Surface	1.98	19.5	37.5	6 30	
			Rottom	19.2	17.4	38.0	14.9	
		10	Middle	9 50	18.8	38.0	15.9	
	-	10	Surface	1.00	20.0	39.0	4 30	
	01		Rottom	18.2	17.3	38.1	23.5	
	2/2	11	Middle	10.2	17.4	38.1	14.9	
	0/	**	Surface	1 42	19.3	36.7	14.9	
ay	-26		Rottom	23.4	17.0	38.1	19.3	
ä	53	12	Middle	13.5	17.0	37.8	3 37	
sta		14	Surface	1.63	19.4	37.0	177	
Ë.			Rottom	29.3	16.6	38.1	3 40	
n l		13	Middle	16.9	17.4	38.1	12.7	
e 7		10	Surface	2 40	19.2	38.1	17.7	
Sid			Rottom	21.10	17.2	38.1	127	
In		17	Middle	11.5	18.0	38.0	129	
			Surface	1 20	19.1	38.0	26.3	
			Rottom	16.5	18.3	38.0	28.2	
		20	Middle	11.3	18.2	37.8	23.5	
		5	Surface	0.50	19.2	38.0	20.3	
			Rottom	20.6	16.9	38.1	20.9 57.8	
		23	Middle	11.2	18.0	38.0	20.6	
			Surface	2.40	19.0	38.0	23.4	
			Bottom	16.3	17.4	38.1	23.0	
		24	Middle	9.40	18.7	38.0	18.7	
			Surface	1.00	19.1	38.0	12.0	
			Bottom	12.7	17.7	38.1	34.9	
		25	Middle	7.30	18.5	38.0	32.0	
			Surface	1.60	19.3	38.0	22.6	
			Bottom	21.0		- 3.0	1.80	
		7	Middle	11.5			9.90	
	12	,	Surface	1 00			<d1< td=""><td></td></d1<>	
	20]		Rottom	26.0			15 7	
)9(15	Middle	20.0 12.5			1.J./ 8.60	
	9/(13	Surface	10.0			6.00	
	3-2		Surjace	1.00			0.00	
	3	01	Bottom	22.0			18.1	
		21	Middle	11.5			14.9	
			Surface	1.00			1.00	

Table 2: Total (THg) and dissolved mercury (DHg) concentration measured in the seawater column, inside and outside the bay.

_	_						
			26	Bottom	8.00	11.1	5.55
	ay			Surface	2.00	4.55	<d.l.< th=""></d.l.<>
	B		a -	Bottom	27.0	6.10	3.55
	sta	12	27	Middle	16.0	4.80	<d.l.< th=""></d.l.<>
	Ba	20]		Surface	2.00	4.55	<d.l.< th=""></d.l.<>
	Au	57		Bottom	81.3	9.15	3.55
	le 7	5(28	Middle	45.3	5.42	<d.1.< th=""></d.1.<>
	sid	0		Surface	6.10	4.75	<d.1.< th=""></d.1.<>
) nt			Bottom	679	12.0	4.15
	\circ		29	Middle	375	6.15	5.55
				Surface	20.0	2.62	<d.l.< td=""></d.l.<>
						Y	
		<u></u>					
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Seawater sampling



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CHR MAN





Station	Regression fit	Φ (µg m ⁻² d ⁻¹)	
▲ S9	Y=4.1ln(x)+14.7; R ² =0.88	23	
XS18	Y=1.6 x+14.6; R ² = 0.97	56	
S22	Y=3.0ln(x)+24.2; R ² = 0.40	8.7	

Station	Regression fit	Φ (µg m ⁻² d ⁻¹)
▲ S7	$Y=20.6ln(x)-19.7; R^2=0.91$	23
OS15	Y=16.6ln(x)+67.8; R ² =0.98	92
¥ S21	Y=3.7x+13.8; R ² =0.92	21

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